

Au-Containing Cuprates

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cuprates

layered structures

high-pressure

superconductivity

1. Introduction

Numerous high-temperature superconducting cuprates have complex perovskite-based intergrowth crystal structures. In these structures, it is possible to select different blocks, for example, perovskite-like (CaTiO_3), halite-like (NaCl), fluorite-like (CaF_2) and/or other ones stacking along the four-fold axis of the perovskite sublattice [\[1\]](#). The first member of series of Ruddlesden-Popper (RP) phases La_2CuO_4 has a simple intergrowth structure consisting of perovskite-like and halite-like fragments (K_2NiF_4 structural type). High-temperature superconductivity at $T_c = 36$ K was discovered in this material after hole doping by Ba^{2+} [\[2\]](#). T_c was shown to increase up to 52.5 K when measured in situ under a hydrostatic pressure of 1.68 GPa [\[3\]](#).

The crystal structures of Hg-containing series of superconducting cuprates, $\text{HgBa}_2\text{Ca}_{n-1}\text{Cu}_n\text{O}_{2n+2+\delta}$ ($n = 1-6$), can be described as intergrowth ones [\[4\]](#)[\[5\]](#)[\[6\]](#)[\[7\]](#)[\[8\]](#)[\[9\]](#). Defect halite-like fragments and perovskite-like fragments are stacking along the four-fold axis, which coincides with the crystallographic c -axis (Figure 1). Hg^{2+} cations have typical dumbbell-like coordination with a significant range of oxygen stoichiometries in the HgO_δ layers. In the crystal structures of compounds with $n = 2-6$, the corresponding amounts of CuO_2 layers are separated by Ca^{2+} cations forming repeated perovskite-like blocks. It is important to note that the highest T_c (up to 138 K) was reported for the compound with $n = 3$. Moreover, at ultrahigh pressures above 20 GPa T_c increased remarkably and reached 166 K in fluorinated Hg-1223 [\[10\]](#)[\[11\]](#)[\[12\]](#).

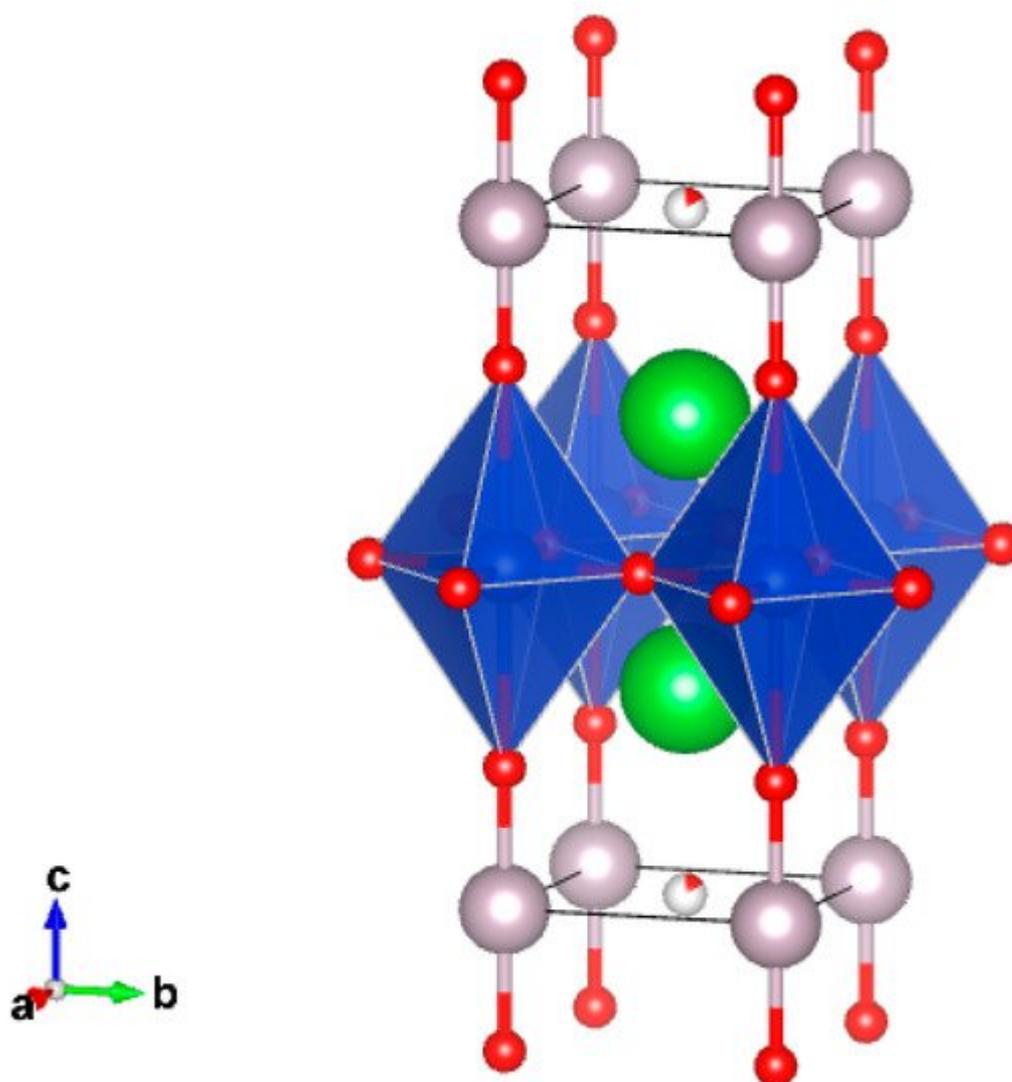


Figure 1. Crystal structure of Hg-1201. The coordination polyhedra of copper (distorted octahedron) and Hg (dumbbell) are shown.

Since the first high-pressure device was invented by Bridgeman in the 1940s, high-pressure has been extensively used in solid state sciences. It is established that ultrahigh pressure deals with the pressures greater than 1.01 GPa, and now the pressure scale is extended up to 500 GPa at temperatures exceeding 3000 °C. The most frequently used apparatus in high pressure research is the diamond-anvil cell (DAC) in a piston-cylinder design. It allows one to achieve pressures as high as 4 GPa or even more. The disadvantage of the Bridgeman device in a piston-cylinder design is the small sample size. That problem was solved by the belt-like apparatus invented in the 1960s, that can generate 10 GPa at temperatures up to 2000 °C. The modern devices are multi-anvil cells, where the anvils are usually made from materials with high hardness, preferably from diamond. Detailed descriptions and a comparison of different high-pressure apparatus are given in a recent review [\[13\]](#).

2. Au-Containing Cuprates: High-Pressure Synthesis and Characterization

High pressure applications is strongly required for the synthesis of these materials because it is only way to stabilize gold oxides which are thermally very unstable at ambient pressures. The general formula of the homologous series is as follows: $\text{AuBa}_2(\text{Ca},\text{Ln})_{n-1}\text{Cu}_n\text{O}_{2n+3}$ ($n = 1-4$; Ln = rare-earth trivalent cations). The second member $\text{AuBa}_2\text{Y}_{1-x}\text{Ca}_x\text{Cu}_2\text{O}_7$ was synthesized in polycrystalline and single crystal form at 1.8 GPa/950 °C and at 6 GPa/1380 °C, respectively [14][15]. Its crystal structure was determined from single crystal X-ray diffraction data in space group Pmmm (Jana 2000 software; $R, R_w (I > 3 \sigma (I)) = 0.0260, 0.0219$; $R, R_w (\text{all data}) = 0.0443, 0.0225$; the details are given in [15]); $a = 3.8260(2) \text{ \AA}$, $b = 3.8501(2) \text{ \AA}$, $c = 12.075 (1) \text{ \AA}$. The samples exhibited bulk superconductivity with $T_c = 82-84 \text{ K}$. The crystal structure of Au-1212 compound is shown in Figure 2.

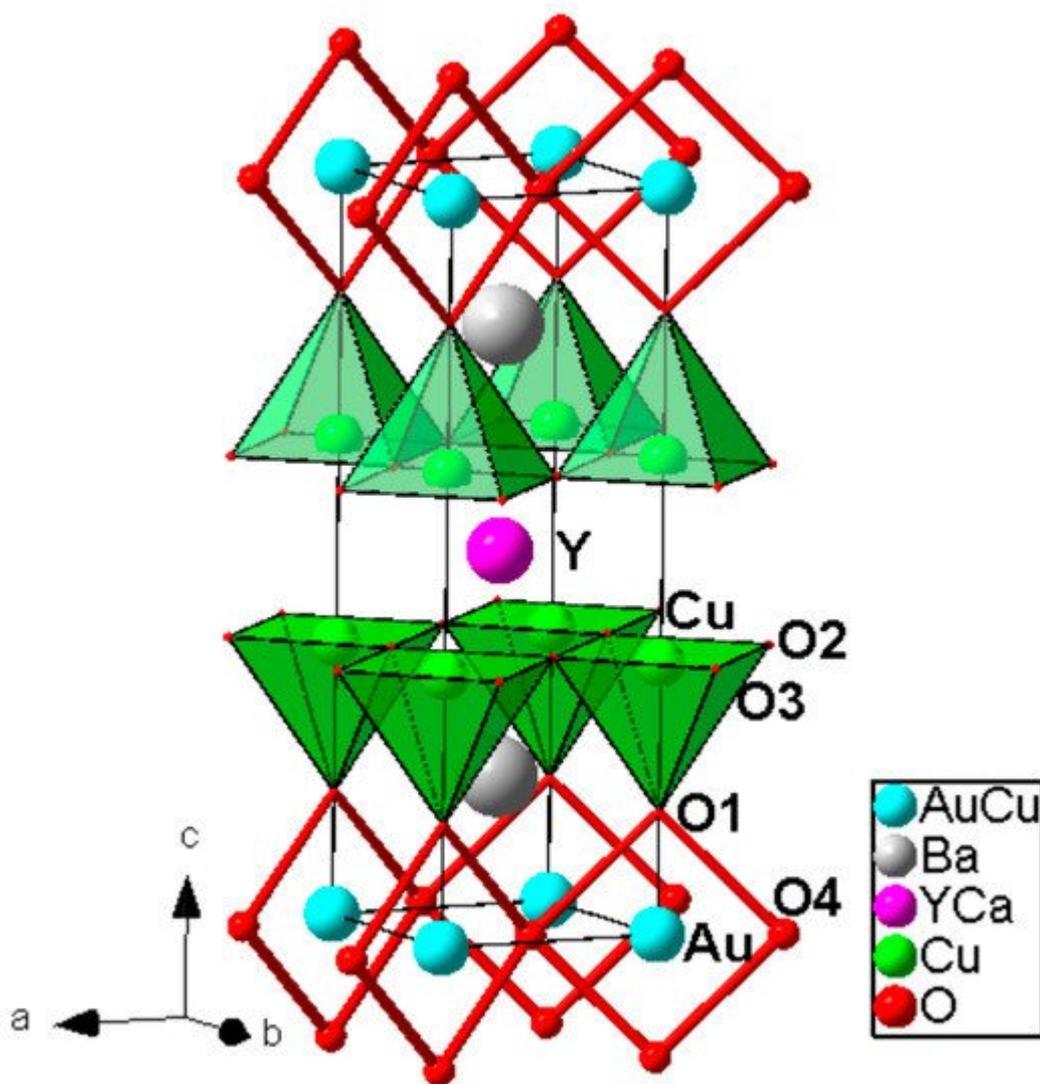


Figure 2. Crystal structure of the Au-1212 compound. Zigzag chain of AuO4 square-planar units is selected. Reproduced with permission from [9].

The cation stoichiometry Au: Ba: Y: Ca: Cu = 0.8(1): 1.9(1): 0.7(1): 0.4(1): 2.2(1) obtained from electron probe microanalysis (EPMA) data is in agreement with that obtained from X-ray single crystal diffraction, indicating ca. 20% substitution of Cu^{1+} cations on the Au^{3+} site, it was associated with approximately the same amount of oxygen

vacancies in the corresponding plane. It should be noted that no superstructure related to possible Cu/Au ordering was revealed from the X-ray single crystal diffraction data.

Au^{3+} is located in square-planar coordination with Au-O distances 1.980 Å and 2.027 Å. AuO_4 squares share corners and they are forming zigzag chains along the short *a*-axis, and that is different from $\text{YBa}_2\text{Cu}_3\text{O}_7$ where CuO_4 squares ran along the longer *b*-axis. The coordination of the in-plane Cu is square pyramidal with an apical distance of 2.41 Å, it is typical for layered cuprate superconductors.

The third and fourth members of the Au-series, $\text{Au}_{1+x}\text{Ba}_2\text{Ca}_2\text{Cu}_{3-x}\text{O}_9$ and $\text{AuBa}_2\text{Ca}_3\text{Cu}_4\text{O}_{11}$, were synthesized at 6 GPa and 1250–1300 °C [16]. They crystallize in primitive orthorhombic unit cells with $a = 3.8182(4)$ Å, $b = 3.8555(4)$ Å, $c = 15.445(2)$ Å and $a = 3.8266(3)$ Å, $b = 3.8505(3)$ Å, $c = 18.494(1)$ Å, for Au-1223 and Au-1234, respectively.

A HRTEM image of Au-1223 phase along the *b*-axis is shown in Figure 3. It is confirmed the 1223 stacking along the *c*-axis: $\text{BaO-AuO-BaO-CuO}_2\text{-Ca-CuO}_2\text{-Ca-CuO}_2$. Attempts to synthesize next members ($n = 5, 6$) were not successful both at 1.8 GPa and at 6 GPa. $\text{AuBa}_2\text{Ca}_3\text{Cu}_4\text{O}_{11}$ showed bulk superconductivity with $T_c \sim 99$ K, but Au-1223 showed only a weak diamagnetic signal at $T_c \sim 30$ K. In order to understand the possible reason, crystal structure of Au-1223 was determined from single crystal X-ray diffraction data (SHELXL-97 software; S.G. Pmmm; $R1, wR2 (I > 2 \sigma (I)) = 0.0188, 0.0437$; R, R_w (all data) = 0.0301, 0.0484; the details are available in [17].

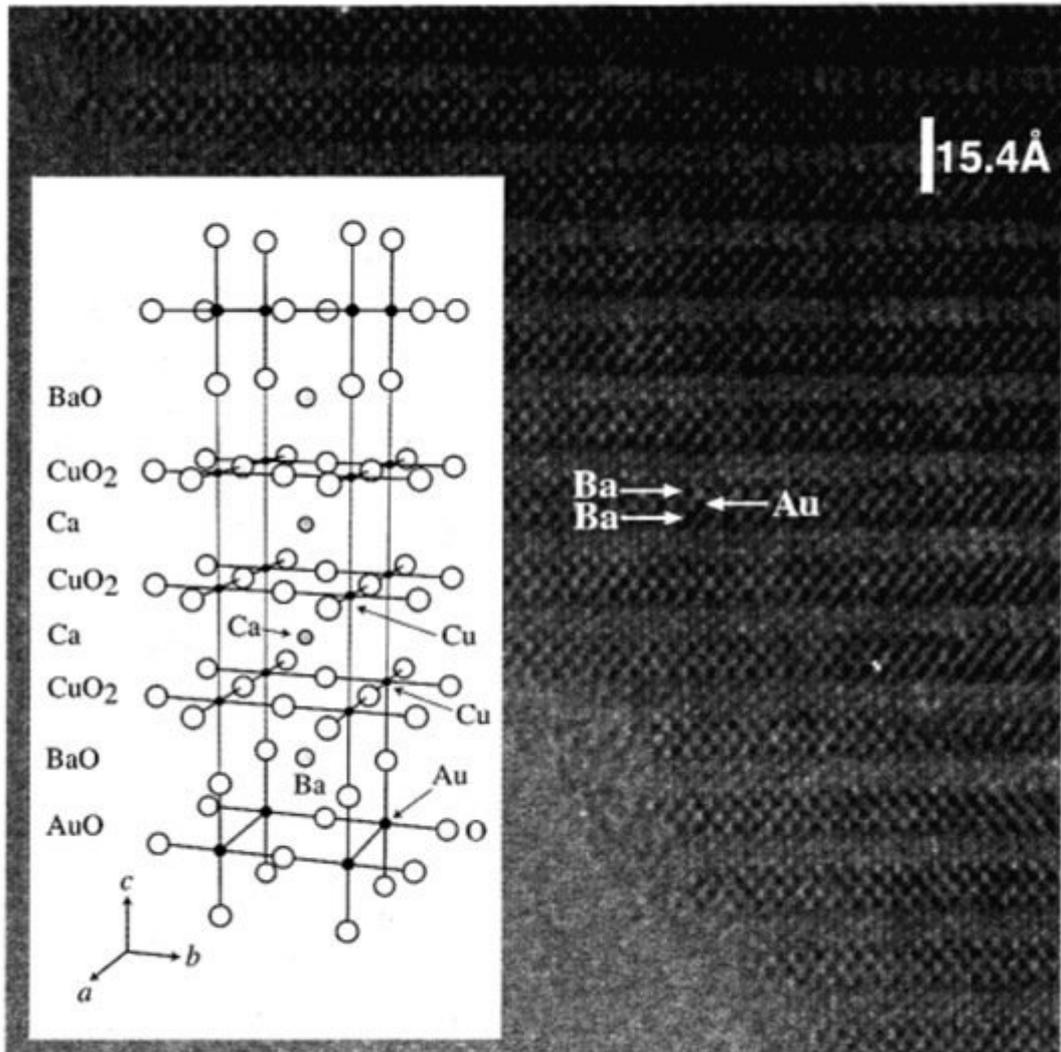


Figure 3. HRTEM image of Au-1223 phase taken along (100) zone axis. The structure model of Au-1223 is shown in the inset. Reproduced with permission from [10].

The cation stoichiometry Au: Ba: Ca: Cu = 0.9(1): 1.91(6): 2.10(6): 3.1(1) was obtained from EPMA. The crystal structure is similar to Hg-1223, but the cation stoichiometry is indicative of some possibility of Cu/Au disorder. The refinement of the occupation yielded 4.1% of Cu in Au site. Au-O in-plane distance is 2.055 Å and it is noticeably longer than typical Cu-O in-plane distances that correlate with their ionic radii difference.

Contrary to Hg-1223, oxygen in the AuO layer is located at $(x, 1/2, 0)$, it leads to formation of AuO₄ zigzag chains as in Au-1212. Electron diffraction revealed the weak superstructure with $b_s = 2b$ in Au-1223, however, the corresponding intensities in X-ray single crystal diffraction data were extremely weak. This suggests no ordering of zigzag-like chains and/or their irregularity.

Related to the suppression of superconductivity, some incorporation of Au for Cu in the CuO₂ planes (6–8%) should be taken into consideration. Although the degree of the substitution is very moderate, even this level can lead to superconductivity suppression.

Au-1201 is the less known member of the Au-series. $\text{Au}(\text{Ba},\text{La})_2\text{CuO}_{5\pm\delta}$ was obtained as a principal phase in the polyphasic sample prepared at 6 GPa and 1250 °C [18]. It crystallizes in an orthorhombic primitive cell with the parameters $a = 3.7976(3) \text{ \AA}$, $b = 3.8509(3) \text{ \AA}$, $c = 8.5749(9) \text{ \AA}$. EPMA shown the cation stoichiometry Au: Ba: La: Cu = 0.75(8): 0.8(1): 1.2(1): 1.25(8), therefore, the approximate formula of Au-1201 can be expressed as $\text{Au}_{0.75}\text{Ba}_{0.8}\text{La}_{1.2}\text{Cu}_{1.25}\text{O}_{5\pm\delta}$. Therefore, it suggests that ca. 25% of Au positions are occupied by Cu. HRTEM indicated that this is a first member of $\text{AuBa}_2(\text{Ca},\text{Ln})_{n-1}\text{Cu}_n\text{O}_{2n+3}$ series (Figure 4). However, an ED study revealed a weak superstructure with $a_s = 5a$, $b_s = b$, $c_s = 2c$. It could be due to possible Au/Cu ordering. Samples of Au-1201 showed a weak diamagnetic signal with $T_c = 19 \text{ K}$ and low level of diamagnetic Meissner fraction.

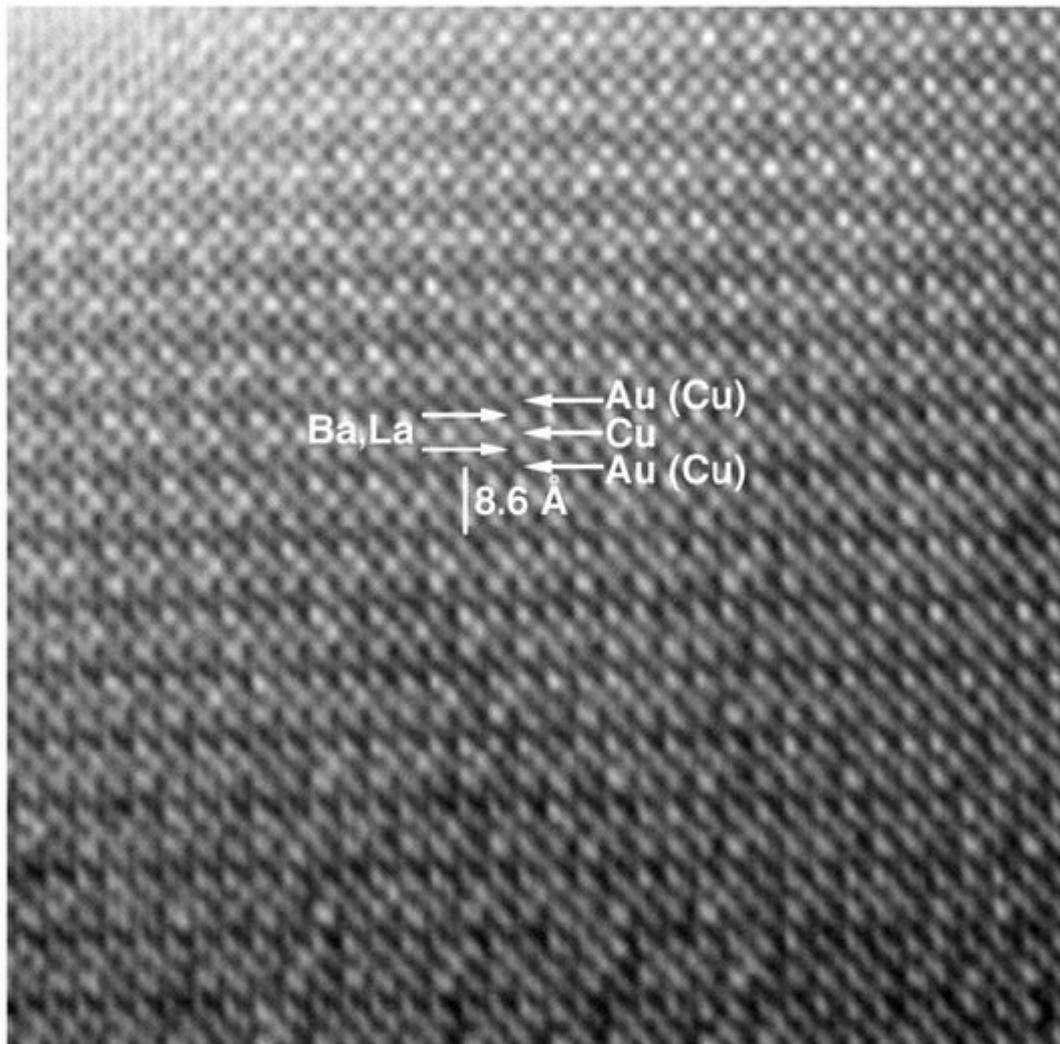


Figure 4. HRTEM image of Au-1201 compounds taken along (010) zone axis. Reproduced with permission from [12].

References

1. Abakumov, A.M.; Antipov, E.V.; Kovba, L.M.; Kopnin, E.M.; Putilin, S.N.; Shpanchenko, R.V. Complex oxides with coherent intergrowth structures. *Russ. Chem. Rev.* 1995, 64, 719–729.
2. Bednorz, J.G.; Müller, K.A. Possible high T_c superconductivity in the Ba-La-Cu-O system. *Z. Phys. B* 1986, 64, 189.
3. Chu, C.W.; Hor, P.H.; Meng, R.L.; Gao, L.; Huang, Z.J. Superconductivity at 52.5 K in the Lanthanum-Barium-Copper-Oxide System. *Science* 1987, 235, 567–569.
4. Putilin, S.N.; Antipov, E.V.; Chmaissem, O.; Marezio, M. Superconductivity at 94 K in $\text{HgBa}_2\text{CuO}_{4+\delta}$. *Nature* 1993, 362, 226.
5. Schilling, A.; Cantoni, M.; Guo, J.D.; Ott, H.R. Superconductivity above 130 K in the Hg–Ba–Ca–Cu–O system. *Nature* 1993, 363, 58.
6. Antipov, E.; Loureiro, S.; Chaillout, C.; Capponi, J.; Bordet, P.; Tholence, J.; Putilin, S.; Marezio, M. The synthesis and characterization of the $\text{HgBa}_2\text{Ca}_2\text{Cu}_3\text{O}_{8+\delta}$ and $\text{HgBa}_2\text{Ca}_3\text{Cu}_4\text{O}_{10+\delta}$ phases. *Phys. C Supercond.* 1993, 215, 1–10.
7. Capponi, J.; Kopnin, E.; Loureiro, S.; Antipov, E.; Gautier, E.; Chaillout, C.; Souletie, B.; Brunner, M.; Tholence, J.; Marezio, M. High-pressure synthesis and heat treatments of the $\text{HgBa}_2\text{Ca}_4\text{Cu}_5\text{O}_{12+\delta}$ and $\text{HgBa}_2\text{Ca}_5\text{Cu}_6\text{O}_{14+\delta}$ phases. *Phys. C Supercond.* 1996, 256, 1–7.
8. Karpinski, J.; Schwer, H.; Conder, K.; Meijer, G.I.; Kopnin, E.; Molinski, R. High pressure crystal growth of $\text{Y}_2\text{Ba}_4\text{Cu}_6+n\text{O}_{14+n}$ and $\text{HgBa}_2\text{Ca}_n-1\text{Cu}_n\text{O}_{2n+2+\delta}$ superconductors. *Solid State Ionics* 1997, 101–103, 985.
9. Schwer, H.; Molinski, R.; Kopnin, E.; Meijer, G.I.; Karpinski, J. Structure and Properties of 1256 and 1267 Type $\text{Hg}_{1-x}\text{R}_x\text{Ba}_2\text{Ca}_n-1\text{Cu}_n\text{O}_{2n+2+4x+d}$ single crystals. *J. Solid State Chem.* 1999, 43, 277.
10. Núñez-Regueiro, M.; Tholence, J.-L.; Antipov, E.V.; Capponi, J.-J.; Marezio, M.; Anderson, P.W. Pressure-Induced Enhancement of T_c Above 150 K in Hg-1223. *Science* 1993, 262, 97–99.
11. Monteverde, M.; Acha, C.; Nuñez-Regueiro, M.; Pavlov, D.A.; Lokshin, K.A.; Putilin, S.N.; Antipov, E.V. High-pressure effects in fluorinated $\text{HgBa}_2\text{Ca}_2\text{Cu}_3\text{O}_{8+\delta}$. *Europhys. Lett.* 2005, 72, 458.
12. Yamamoto, A.; Takeshita, N.; Terakura, C.; Tokura, Y. High-pressure effects revisited for the cuprate superconductor family with highest critical temperature. *Nat. Commun.* 2015, 6, 8990.
13. Liu, X.Y. High pressure synthesis and preparation of inorganic materials. In *Modern Inorganic Synthetic Chemistry*, 2nd ed.; Elsevier: Amsterdam, The Netherlands, 2017; Chapter 5; pp. 105–141.
14. Bordet, P.; Lefloch, S.; Chaillout, C.; Duc, F.; Gorius, M.; Perroux, M.; Capponi, J.; Toulemonde, P.; Tholence, J. $\text{AuBa}_2(\text{Y}_{1-x}, \text{Ca}_x)\text{Cu}_2\text{O}_7$: A new superconducting gold cuprate with T_c above 80 K. *Phys. C Supercond.* 1997, 276, 237–244.

15. Bordet, P.; Kopnin, E.M.; Sato, A.; Takayama-Muromachi, E. Structure analysis of superconducting Au-1212 cuprate. *Supercond. Sci. Technol.* 2003, 16, 685–689.
16. Kopnin, E.M.; Loureiro, S.M.; Asaka, T.; Anan, Y.; Matsui, Y.; Takayama-Muromachi, E. New family of Au-based superconductors $\text{AuBa}_2\text{Ca}_n\text{Cu}_{n-1}\text{O}_{2n+3}$ ($n = 3, 4$). *Chem. Mater.* 2001, 13, 2905.
17. Kopnin, E.; Sato, A.; Asaka, T.; Matsui, Y.; Takayama-Muromachi, E. Structure analysis of Au-containing cuprate of $\text{Au}_{1+x}\text{Ba}_2\text{Ca}_2\text{Cu}_{3-x}\text{O}_9$ (Au-1223). *Phys. C Supercond.* 2003, 387, 406–410.
18. Kopnin, E.; Sato, A.; Asaka, T.; Matsui, Y.; Takayama-Muromachi, E. High-pressure synthesis and characterization of the Au-1201 phase. *J. Alloy. Compd.* 2003, 361, 28–31.

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